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### DYNAMIC SIMULATION AND COMPOSITION CONTROL IN A 10 L MIXING TANK

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#### Abstract

The open loop experiment of composition dynamic in a 10 L mixing tank has been successfully done in laboratory. A 10 L tank was designed for mixing of water (as a stream-1) and salt solution (as a stream-2 with salt concentration,  $c_2$  constant). An electric stirrer was employed to obtain uniform composition in tank. In order to keep the liquid volume constant, the system was designed overflow. In this work, 2 composition control configurations have been proposed; they are Alternative-1 and Alternative-2. For Alternative-1, the volumetric-rate of stream-1 was chosen as a manipulated variable, while the volumetric-rate of stream-2 was chosen as a manipulated variable for Alternative-2. The composition control parameters for both alternatives have been tuned experimentally. The volumetric-rate of manipulated variable was changed based on step function. The outlet stream's composition response  $(c_3)$  to a change in the input volumetric-rate has been investigated. This experiment gave Proportional Integral Derivative (PID) control parameters. The gain controllers  $K_c$ [cm<sup>6</sup>/(gr.sec)] for Alternative-1 and Alternative-2 are -34200 and 40459 respectively. Integral time constant ( $\tau_{\rm I}$ ) and Derivative time constant ( $\tau_{\rm D}$ ) for both alternatives are the same, i.e.  $\tau_{\rm I} = 16$  second, and  $\tau_D = 4$  second. Furthermore, closed loop dynamic simulation using computer programming was also done to evaluate the resulted tuning parameters. The developed mathematical model of composition control system in a mixing tank was solved numerically. Such mathematical model was rigorously examined in Scilab software environment. The results showed that closed loop responses in PID control were faster than those in P and PI controls.

Keywords: closed loop; mixing tank; open loop; pid control; step function

#### Abstrak

DINAMIKA SIMULASI DAN PENGENDALIAN KOMPOSISI DALAM TANGKI PENCAMPUR BERKAPASITAS 10 L. Percobaan loop terbuka dinamika komposisi dalam tangki pencampur 10 L telah dilaksanakan di laboratorium. Tangki 10 L dirancang untuk proses pencampuran air (sebagai arus-1) dan larutan garam (sebagai arus-2 dengan konsentrasi garam, c<sub>2</sub> konstan). Pengadukan diterapkan untuk mencapai keseragaman komposisi di dalam tangki. Untuk menjaga volume cairan di dalam tangki konstan, sistem dirancang overflow. Penelitian ini mengusulkan 2 konfigurasi pengendalian komposisi, yaitu Alternatif-1 dan Alternatif-2. Untuk Alternatif-1, laju alir volumetrik arus-1 dipilih sebagai variabel yang dimanipulasi (MV), sedangkan arus-2 dipilih sebagai MV untuk Alternatif-2. Parameter-parameter pengendalian komposisi untuk kedua alternatif ditentukan melalui percobaan. Laju alir volumetrik dari MV diubah mengikuti fungsi tahap. Percobaan dilanjutkan dengan pengamatan respons komposisi keluaran tangki (c3) terhadap perubahan input fungsi tahap. Percobaan ini menghasilkan parameter pengendalian PID (Proportional Integral Derivative). Gain pengendali  $K_c$  [cm<sup>6</sup>/(gr.sec)] untuk Alternatif-1 and Alternatif-2 berturut-turut adalah -34200 dan 40459. Konstanta waktu integral ( $\tau_i$ ) dan konstanta waktu derivative ( $\tau_D$ ) untuk kedua alternatif adalah sama, yaitu  $\tau_I = 16$  detik, and  $\tau_D = 4$  detik. Kemudian, simulasi loop tertutup menggunakan pemrograman komputer dilaksanakan untuk mengevaluasi parameter pengendalian PID yang dihasilkan. Model matematika sistem pengendalian komposisi dalam tangki pencampur diselesaikan secara numerik dan diuji menggunakan software Scilab. Hasil simulasi menunjukkan bahwa respons loop tertutup pengendalian PID lebih cepat dibandingkan respons pengendalian P dan PI.

Kata kunci: loop tertutup; tangki pencampur; loop terbuka; pengendalian pid; fungsi tahap

#### INTRODUCTION

A mixing tank is frequently used in chemical process industries, for examples as a blending tank and/or a continuous stirred tank reactor. Liquid composition in a mixing tank is one of important parameters for mixing processes or chemical reaction processes in reactor. The propagation of mass disturbance is possibly occurred in mixing processes. Therefore composition control should be implemented to overcome the propagation of mass disturbances.

Composition control parameters such as proportional gain controller (K<sub>c</sub>), integral time constant ( $\tau_l$ ), and derivative time constant ( $\tau_D$ ) should be tuned properly, since they potentially affect the stability of mixing process. However, designed composition control system must be able to give a stable response in facing the mass disturbances. Therefore the study on the dynamic simulation and composition control is very important.

Some studies on process dynamic and control have been done previously. Recently, Hermawan *et al.* (2012) have presented the open loop composition dynamic in a 10 L Mixing Tank experimentally. Hermawan *et al.* (2010) have also presented the design of control configuration of non-interacting-tank system using quantitative analysis of relative gain array. Hermawan (2011) has implemented Process Reaction Curve (PRC) for tuning of temperature control parameters in a 10 L Stirred Tank Heater. Widayati and Hermawan (2007) have studied the mixing characteristic in a horizontal stirred tank.

The goals of this research were to propose the composition control configuration and to tune the composition control parameters (PID Control parameters) in a 10 L Mixing Tank. The resulted composition control parameters of proposed configurations were then examined through dynamic simulation. In order to achieve the aims of this research, this work was done in two parts, i.e. open

loop experiment in the laboratory for tuning of composition control parameters and closed loop simulation using computer programming to explore dynamic behavior of the controlled system. The open loop experiment in the laboratory was carried out to tune composition control parameters. The volumetric rate of input stream was chosen as a manipulated variable to maintain the concentration of output stream at a constant value. In order to examine the control configuration, the mass disturbances were made based on step function. The Scilab software was utilized to carry out dynamic simulation.

#### MATERIALS AND METHODS

The experimental apparatus setup is shown in Figure 1. As can be seen in Figure 1, No.1 is a main tank that represents a mixing tank. This mixing tank has 2 input streams, i.e. stream-1 and stream-2, and an output stream, i.e. stream-3. In normal condition, stream-1 and stream-2 come from the feeding tank No. 2 and No. 3, respectively. In this work, water was used as a stream-1 with its volumetric rate f1 [cm<sup>3</sup>/sec], and salt solution as a stream-2 with its volumetric rate  $f_2$  [cm<sup>3</sup>/sec] and concentration  $c_2$  $[gr/cm^3]$ . The input concentration  $c_2$  is constant. The output stream (stream-3) has volumetric rate f3  $[cm^{3}/sec]$  and concentration  $c_{3}$  [gr/cm<sup>3</sup>]. The concentration c3 is measured by means of Conductivity-meter. Since the liquid volume is kept constant, the system is designed overflow. A stirrer is employed to obtain uniform composition in the mixing tank. Tank No.4 was used if we want to make a concentration disturbance of stream-2. This disturbance could be made by changing the inlet concentration of stream-2 c2 immediately. This is done by revolving the gate of three-way-valve (No. 7 in Figure 1), so that stream-2 comes from the tank No. 4 which is specifically prepared for making disturbance.

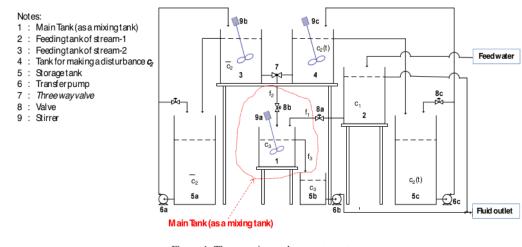


Figure 1. The experimental apparatus setup

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The material balance of the mixing tank can be written as follows:

$$\frac{\mathrm{d}\mathbf{c}_{3}(\mathbf{t})}{\mathrm{d}\mathbf{t}} = \left[f_{1}(\mathbf{t})\overline{\mathbf{c}}_{1} + f_{2}(\mathbf{t})\overline{\mathbf{c}}_{2} - (f_{1}(\mathbf{t}) + f_{2}(\mathbf{t}))\mathbf{c}_{3}(\mathbf{t})\right]/\mathbf{V} \qquad (1)$$

In this research, two composition control configurations are proposed, i.e. Alternative-1 and Alternative-2 as shown in Figure 2. Open loop tuning experiment is done for either alternatives by changing the opening valve of stream-1 (No. 8a in Figure 1) or stream-2 (No. 8b in Figure 1) to increase/decrease its volumetric rate immediately. The output concentration (c<sub>3</sub>) response to a change in input volumetric rate was then investigated. The resulted response will similar with that response given by first order plus dead time (FOPDT) model. PID Control parameters were then tuned by fitting the resulted FOPDT as proposed by Ziegler-Nichols (Smith and Corripio, 1997). These open loop experiments should be started from its initial (normal) conditions.

In order to evaluate the resulted PID Control parameters, dynamic simulation was carried out by means of a computer. A simple feedback control system was implemented to maintain liquid concentration in tank (c<sub>3</sub>) constant by manipulating the volumetric rate of stream-1 or stream-2. Thus, the equation of manipulated variables for both of control configuration alternatives can be written as follow: Alternative-1:

$$f_1(t) = \overline{f}_1 + K_c e(t) + \frac{K_c}{\tau_1} \int e(t) dt + K_c \tau_D \frac{de(t)}{dt}$$
(2)

Alternative-2:

$$f_2(t) = \overline{f}_2 + K_c e(t) + \frac{K_c}{\tau_1} \int e(t) dt + K_c \tau_D \frac{de(t)}{dt}$$
(3)

Where e(t) is defined as:

$$e(t) = c_3^{SP} - c_3(t) = error$$
(4)

The developed mathematical model of composition control system in the mixing tank was solved numerically with the easiest way, that was Explicit Euler. The free software Scilab was chosen to carry out the closed loop dynamic simulation. The closed loop responses of composition control could then be explored in this work.

#### RESULTS AND DISCUSSION

Steady state parameters of the mixing tank are listed in Table 1. Based on steady state material balance, the process time constant is found to be 37 seconds (0.6 minutes). Therefore the system is considered quite sensitive to the changes of input disturbances.

## Tuning of Composition Control Parameters for Alternative-1

For Alternative-1, volumetric rate of water ( $f_1$ ) is considered as a manipulated variable to maintain liquid composition in tank ( $c_3$ ). Figure 3.a shows the influence of  $f_1$  on  $c_3$ . Volumetric rate of water decreases by an amount of 76 cm<sup>3</sup>/sec immediately; the concentration  $c_3$  rises about 0.01 gr/cm<sup>3</sup>. The tuning results of composition control parameters (P, PI, and PID) for Alternative-1 are listed in Table 2.

### Tuning of Composition Control Parameters for Alternative-2

For Alternative-2, volumetric rate of salt solution ( $f_2$ ) is considered as a manipulated variable to maintain liquid composition in tank ( $c_3$ ). Figure 3.b shows the open loop composition response to a change in the volumetric rate  $f_2$ . The concentration  $c_3$  increases (about 0.01 gr/cm<sup>3</sup>) as the volumetric rate  $f_2$  increases (about 70 cm<sup>3</sup>/sec). The tuning results of composition control parameters (P, PI, and PID) for Alternative-2 are also listed in Table 2.

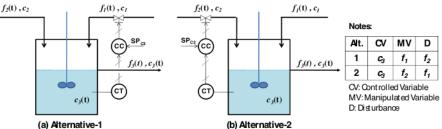


Figure 2. Composition Control Configuration: (a) Alternative-1, (b) Alternative-2.

No	Variable	Steady stat	
1	Volumetric rate of stream-1, $f_1$ (cm <sup>3</sup> /second)	106	
2	Volumetric rate of stream-2, $f_2$ (cm <sup>3</sup> /second)	71	
3	Volumetric rate of stream-3, $f_3$ (cm <sup>3</sup> /second)	177	
4	Concentration of stream-1, $c_1$ (gr/cm <sup>3</sup> )	0	
5	Concentration of stream-2, $c_2$ (gr/cm <sup>3</sup> )	0.05	
6	Concentration of stream-3, $c_3$ (gr/cm <sup>3</sup> )	0.0214	
7	Liquid volume in tank, $V(\text{cm}^3)$	6600	

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#### (Hermawan and Haryono)

Type of Feedback Control	Proportional Gain Kc [cm <sup>6</sup> /(gr.sec)]			s of composition control parameter Integral time $\tau_I$ [sec]			ers Derivative time $\tau_D$ [sec]		
	Kc	Alt-1	Alt-2	$\tau_I$	Alt-1	Alt-2	$\tau_D$	Alt-1	Alt-2
Р	$\tau / (K t_D)$	-28500	33716	-	-	-	-	-	-
PI	$0.9 \tau / (K t_D)$	-25650	30344	3.3 t <sub>p</sub>	27	27	-	-	-
PID	$1.2 \tau / (K.t_D)$	-34200	40459	$2t_D$	16	16	$0.5 t_D$	4	4

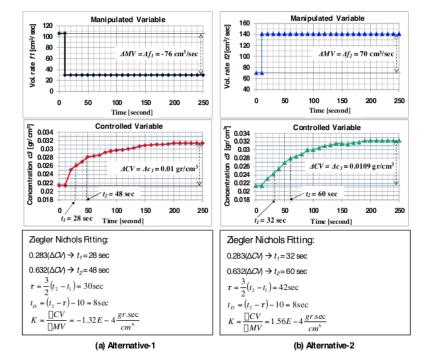


Figure 3. Tuning of Composition Control Parameters: (a) Alternative-1, (b) Alternative-2.

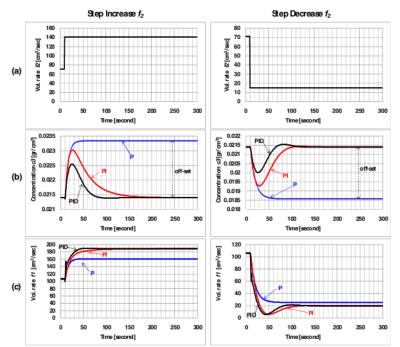
#### Dynamic Simulation of Composition Control for Alternative-1

Closed loop responses to a change in volumetric rate f2 are illustrated in Figure 4. The disturbances were made by following both functions of step increase and step decrease. For step increase's disturbance, volumetric rate f2 is increased by an amount of 70 cm3/sec at time equals 10 seconds. As can be seen, the composition controller (P, PI, and PID) attempts to return concentration c3 to its normal value of 0.0214 gr/cm3. P Control produces an offset of 0.0019 gr/cm3. Combination of proportional and integral control modes leads to eliminate an offset. Concentration c<sub>3</sub> can be returned to its set point by both of PI and PID Controls. Closed loop response of PID Control is fastest compared to P and PI Controls; Concentration  $c_3$  can be returned to its set point at time equals 150 seconds.

For step decrease's disturbance, volumetric rate  $f_2$  is decreased by an amount of 56 cm<sup>3</sup>/sec at time

equals 10 seconds. The concentration  $c_3$  decreases first, and then rises to its normal value. However P Control still produces an off-set of about 0.0028 gr/cm<sup>3</sup>. Closed loop response of PID Control is the fastest; the set point of  $c_3$  can be achieved at time equals 120 sec.

Combination of the three control modes (PID control) gives a closed loop response which has in general the same qualitative dynamic characteristics as those resulting from PI control. To increase the speed of the closed loop response we can increase the value of proportional gain (K<sub>e</sub>) and/or decrease the value of integral time constant ( $\tau_l$ ). But increasing K<sub>e</sub> and/or decreasing  $\tau_l$ , the response become more oscillatory and may lead to instability. The introduction of the derivative mode brings a stabilizing effect to the system. Therefore, the derivative control action not only produces faster response but also produce more robust response.



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Figure 4. Closed Loop Responses of Composition Control Alternative-1 to a change in volumetric rate  $f_2$ : (a) volumetric rate  $f_2$ , (b) concentration  $c_3$ , (c) volumetric rate  $f_1$ 

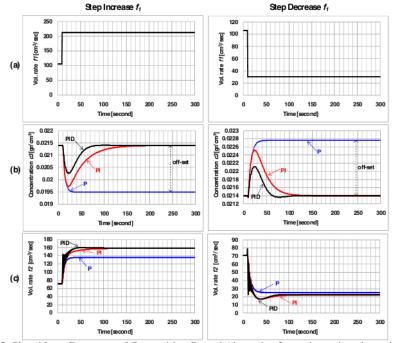


Figure 5. Closed Loop Responses of Composition Control Alternative-2 to a change in volumetric rate  $f_1$ : (a) volumetric rate  $f_1$ , (b) concentration  $c_3$ , (c) volumetric rate  $f_2$ 

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#### Dynamic Simulation of Composition Control for Alternative-2

Figure 5 shows closed loop responses to a change in volumetric rate  $f_1$ . For this alternative, the disturbances were also made by following both functions of step increase and step decrease. For step increase's disturbance, volumetric rate f1 is increased by an amount of 106 cm3/sec at time equals 10 seconds. As shown in Figure 5, concentration c3 decreases as volumetric rate f1 increases, and then concentration c3 can be returned to its set point by both of PI and PID Controls. P Control produces an offset of 0.0019 gr/cm3. Closed loop response of PID Control is the fastest one; concentration c3 can be returned to its set point at time equals 150 seconds.

For step decrease's disturbance, volumetric rate f<sub>1</sub> is decreased by an amount of 76 cm<sup>3</sup>/sec at time equals 10 seconds. The concentration c3 increases as the volumetric rate of water decreases, and then drops to its normal value for PI and PID Controls. Again, P Control still produces an off-set of about 0.0014 gr/cm<sup>3</sup>, and PID Control gives the fastest response.

#### CONCLUSIONS

This paper discussed tuning of composition control parameters and dynamic simulation in a 10 L mixing tank. Two alternatives of composition control configurations have been proposed. Closed loop dynamic behaviours of the two control configurations have also been explored. According to the dynamic simulation, the tuning results of composition control parameters produce stable responses. This research reveals that PID Composition Control produces the fastest responses compared to both of P and PI Composition Controls.

#### ACKNOWLEDGEMENTS

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#### NOMENCLATURE

- steady state change in controlled variable //CV [gr/cm<sup>3</sup>]
- $\Box MV$ step change in manipulated variable [cm<sup>3</sup>/second]
- effective process time constant [second]  $\tau$
- derivative time constant [second]  $\tau_D$

#### (Hermawan and Haryono)

- integral time constant [second]  $\tau_l$
- $c_{I,2,3}_{SP}$ concentration of stream 1, 2, 3 [gr/cm<sup>3</sup>] set point of liquid concentration in tank Ci
- [gr/cm<sup>3</sup>] e
- error [gr/cm3]
- f<sub>1,2,3</sub> volumetric rate of stream 1, 2, 3 [cm<sup>3</sup>/second]
- steady state gain of the process Κ [(gr.second)/cm6]
- proportional gain controller [cm<sup>6</sup>/(gr.second)]  $K_c$
- time at which  $c_3 = 0.283 \square c_{3s}$  [second]  $t_I$
- time at which  $c_3 = 0.632 \prod c_{33}$  [second]  $t_2$
- effective process dead time [second]  $t_D$
- liquid volume in tank [cm3]

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